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Form Approved
OMB No. 0704-0188

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden. to Washington Headquarters Services, Directorate for information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project (0704-0188), Washington, DC 20503.

1. AGENCY USE ONLY (Leave blank)

2. REPORT DATE

January 17, 1995

3. REPORT TYPE AND DATES COVERED

4. TITLE AND SUBTITLE

Synthesis of Tungsten Nitrene Complexes as Precursors to Tungsten Nitride

5. FUNDING NUMBERS

6. AUTHOR(S)

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7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)

HERORT ON

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8. PERFORMING ORGANIZATION REPORT NUMBER

9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)

U.S. Army Research Office

P.O. Box 12211

Research Triangle Park, NC 27709-2211

10. SPONSORING / MONITORING AGENCY REPORT NUMBER

11. SUPPLEMENTARY NOTES

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12a, DISTRIBUTION / AVAILABILITY STATEMENT

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12b. DISTRIBUTION CODE

#### 13. ABSTRACT (Maximum 200 words)

Chemical vapor deposition using organometallic precursors (MOCVD) provides a method for the preparation of thin films. Low valent tungsten nitrene complexes were synthesized as potential precursors to tungsten nitride (WN<sub>x</sub>), a material used in diffusion barriers for Si or GaAs semiconductor devices. The original target precursors for MOCVD of WN<sub>x</sub> were the carbonyl-containing complexes  $(CO)_{5-n}(PR_3)_nW=NR$ , where R is an alkyl or aryl group. Later synthetic work involved the tungsten(IV) imido (or nitrene) complexes  $(CO)_2I_2LW=NPh$ , which were prepared by oxidation of the zwitterionic species  $(CO)_5WNPhNPhC(OMe)Ph$  with one equivalent of  $I_2$  followed by addition of a coordinating species L [L= THF, pyridine, PMe<sub>3</sub>, P(OMe)<sub>3</sub>].

# DITC QUALITY INSPECTED &

1	4. SUBJECT TERMS				_		15. NUMBER OF PAGES 2
1	nitrene complex, in	nido	complex,	tungsten	nitride,	MOCVD precursors	16. PRICE CODE
1	7. SECURITY CLASSIFICATION OF REPORT		SECURITY CLA OF THIS PAGE		19. SECURI OF ABS	TY CLASSIFICATION STRACT	20. LIMITATION OF ABSTRACT
	UNCLASSIFIED	UNCLASSIFIED		UNCL	ASSIFIED	UL	

## **Final Report**

1.	ARO PROPOSAL NUMBE	R: 33087-CH		-		
2.	PERIOD COVERED BY RE	EPORT: 15 May 1994 - 14 Novem	iber 199	)4	A	
3.	TITLE OF PROPOSAL:	Synthesis of Tungsten Nitrene Complexes as Precursors to Tungsten Nitride				
4.	GRANT NUMBER:	DAAH04-94-G-0096	Ву			
5.	NAME OF INSTITUTION:	University of Florida  Distribution /  Availability Co				
6.	AUTHOR OF REPORT:	Lisa McElwee-White	Dist Avail and Special			

### STATEMENT OF PROBLEM:

Chemical vapor deposition using organometallic precursors (MOCVD) provides a method for the preparation of thin films. We are developing syntheses of low valent tungsten nitrene complexes as precursors to tungsten nitride (WN<sub>x</sub>), a material used in diffusion barriers for Si or GaAs semiconductor devices. Our target precursors for MOCVD of WN<sub>x</sub> are the carbonyl-containing complexes (CO)<sub>5-n</sub>(PR<sub>3</sub>)<sub>n</sub>W=NR, where R is an alkyl or aryl group. Upon heating at a surface, these compounds should lose the five ancillary ligands and undergo fragmentation of the NR group to deposit a film of WN<sub>x</sub>.

# 8. SUMMARY OF MOST IMPORTANT RESULTS:

We have recently begun to explore the chemistry of tungsten (IV) imido complexes bearing CO ligands. The new target complexes were molecules of the type (CO)2I2LW≡NPh. These species maintain the CO ligands that should be good leaving groups during MOCVD of tungsten nitride but the higher oxidation state renders the metal nitrene (or imido) moiety much more stable than in the zero-valent nitrene complexes studied in the earlier stages of the project. We had previously reported that the zwitterionic complex (CO)5WNPhNPhC(OMe)Ph decomposes upon thermolysis or photolysis to yield (CO)5W=NPh. We have now demonstrated that this species is also a precursor to the tungsten (IV) imido complexes since the zwitterion can serve as a protected imido functionality during oxidation of the metal center. Treatment of the zwitterionic complex (CO)5WNPhNPhC(OMe)Ph with one equivalent of I2 leads to the formation of highly unusual iodo-bridged tungsten (IV) imido dimer [(CO)<sub>2</sub>I<sub>2</sub>W≡NPh]<sub>2</sub> and phonic (OMe)Ph (Scheme 1). Cleavage of the dimer with coordinating species leads to formation of the monomeric compounds (CO)<sub>2</sub>I<sub>2</sub>LW=NPh [L = pyridine, amines, phosphines, or phosphites] These complexes are now being explored as MOCVD precursors in collaboration with Professor Timothy J. Anderson, Department of Chemical Engineering, University of Florida.

### Scheme 1

### 9. LIST OF PUBLICATIONS:

- "Reactions of (CO)<sub>5</sub>W(THF) with Triphenylmethyl Azide and Triptycyl Azide," Massey, S.T.; Mansour, B.; McElwee-White, L. J. Organomet. Chem., in press.
- 2. "Tungsten (IV) Imido Complexes From Oxidation of a Protected Zero-Valent Nitrene Precursor," McGowan, P.C.; Massey, S.T.; Abboud, K.A.; McElwee-White, L. J. Am. Chem. Soc., 1994, 116, 7419-7420.

# 10. PARTICIPATING SCIENTIFIC PERSONNEL:

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Graduate Students and Postdoctorals received stipends, undergraduates received only research supplies.